## Alkynylation of Thiolactams. New Synthesis of $\alpha$ -Substituted Pyrrolidine and Piperidine Alkaloids

Hiroki Takahata,\* Koichi Takahashi,† Eng-Chi Wang, and Takao Yamazaki

Faculty of Pharmaceutical Sciences, Toyama Medical & Pharmaceutical University, 2630 Sugitani, Toyama 930-01, Japan

Alkynylation of S-alkylthioamidium salts of thiolactams with lithium acetylides followed by reduction with LiAlH<sub>4</sub> provided  $\alpha$ -alkynylazacycloalkanes, which on reduction, or hydroboration followed by oxidation, gave  $\alpha$ -substituted pyrrolidine and piperidine alkaloids.

The development of functionalization methods for azacycloalkanes such as pyrrolidines, piperidines, and perhydroazepines is of great importance in the synthesis of naturally occurring alkaloids.<sup>1</sup> In connection with our studies on the utilization of the thioamide group in organic synthesis,<sup>2</sup> we describe the alkynylation of thiolactams, and subsequent manipulation leading to  $\alpha$ -substituted pyrrolidine and piperidine alkaloids.

The thioamide group has increasingly been recognized as a useful synthon in organic synthesis. Recently, the attack of organolithium reagents upon the carbon-sulphur double bond of thioamides has been explored.<sup>3</sup> Unfortunately, initial attempts at alkynylation of pyrrolidine-2-thione by treatment with lithium acetylides were unsuccessful. Previously, we described the carbon-carbon bond forming reaction of *S*-alkylthioamidium salts with malononitrile.<sup>4</sup> Here we describe the reaction <sup>5</sup> of *S*-alkylthioamidium salts, prepared from thiolactams by treatment with methyl iodide, with lithium acetylide, followed by reduction with lithium aluminium hydride (LiAlH<sub>4</sub>) to give  $\alpha$ -alkynylazacycloalkanes.

Thus, S-methylthioamidium salts (I), generated *in situ* from thiolactams (1)—(3) and methyl iodide in tetrahydrofuran (THF) at room temperature for 8 h, were treated with a lithium acetylide (6)—(8) at -78 °C. The resulting adducts (II) were then treated with LiAlH<sub>4</sub> at -78 °C followed by warming to -20 °C to give the  $\alpha$ -alkynylazacycloalkanes (10)—(17) in moderate yields (Table, entries 1—8). Similarly, the  $\alpha$ -alkynylazacycloalkanes (18)—(22) were obtained from N-benzylthiolactams (4) and (5) (Table, entries 9—13).

Our attention was focussed on the synthesis of pyrrolidine and piperidine alkaloids by the elaboration (reduction or hydroboration) of the acetylene function. Hydrogenation of piperidinylacetylene (17) in the presence of palladium hydroxide  $[Pd(OH)_2]$  in methanol afforded  $(\pm)$ -*N*-methylconiine<sup>‡</sup> (23) quantitatively. Under the same conditions the *N*-benzyl derivative (21) underwent both reduction and hydrogenolysis<sup>6</sup> to provide  $(\pm)$ -coniine (24)<sup>‡</sup> quantitatively.

Hydroboration-oxidation of acetylenes is an effective procedure for regioselective ketone synthesis. However, to our knowledge, the hydroboration of acetylenic amines has scarcely been studied.<sup>7</sup> Monohydroboration of piperidinylacetylene (17) with bis(1,2-dimethylpropyl)borane § in THF at 0 °C followed by oxidative work-up proceeded in a regioselective manner,



Table. Isolated yields of  $\alpha$ -alkynylazacycloalkanes from thiolactams (three-steps sequence)

					Yield
Entry	Product	<b>R</b> <sup>1</sup>	R <sup>2</sup>	n	(%)
1	(10)	Me	Ph	1	70
2	(11)	Me	Ph	2	67
3	(12)	Me	Ph	3	80
4	(13)	Me	$n-C_5H_{11}$	1	71
5	(14)	Me	n-C <sub>5</sub> H <sub>11</sub>	2	74
6	(15)	Me	n-C,H,	3	55
7	(16)	Me	Me	1	61
8	(17)	Me	Me	2	85
9	(18)	CH <sub>2</sub> Ph	Ph	1	78
10	(19)	$CH_2Ph$	$n-C_5H_{11}$	1	67
11	(20)	CH <sub>2</sub> Ph	Me	1	66
12	(21)	$CH_2Ph$	Me	2	65
13	(22)	$CH_2Ph$	TMS	1	80

with ketone formation at the less hindered carbon atom of the triple bond to give  $(\pm)$ -*N*-methylpelletierine (**25**) as a major product (50% yield), and its regioisomer (**26**) (14%). However, the regioselectivity (3.5:1) is not high compared with that (91:9) for the hydroboration of cyclohexylethyne.<sup>8</sup> This somewhat low regioselectivity may be due to co-ordination of the boron of the bis(1,2-dimethylpropyl)borane with the amine of the piperidine

<sup>†</sup> Visiting scientist from Tateyama Kasei Co. Ltd., Kosugi, Toyama, Japan, 1987—1988.

<sup>‡</sup> Melting points of derivatives of these compounds were consistent with those described in the literature; see Experimental section.

<sup>§</sup> Although treatment of (17) with 9-borabicyclo[3.3.1]nonane caused no reaction at room temperature, refluxing in THF provided (25) (34%) and (26) (33%) with no regioselectivity.





as in (27). Similarly,  $(\pm)$ -sedaminone (28)\* was obtained from (11) (50% yield), together with its regioisomer (29) (18%). On the other hand, hydroboration of the pyrrolidinylacetylene (16) showed higher regioselectivity than the piperidinyl analogue (17), providing  $(\pm)$ -hygrine (30) (54%) on oxidation. Similarly, the reaction of (20) proceeded regiospecifically to afford only (31), in 52% yield. This suggested that co-ordination of the boron of bis(1,2-dimethylpropyl)borane with the pyrrolidine nitrogen was difficult, presumably due to steric hindrance between the nitrogen substituent and the dialkyl group of the diborane.

In summary, the present study provides a convenient and simple procedure for the synthesis of  $\alpha$ -alkynylazacycloalkanes, which are transformed into  $\alpha$ -substituted pyrrolidine and piperidine alkaloids. This method should be applicable to the synthesis of many kinds of  $\alpha, \alpha'$ -disubstituted pyrrolidine and piperidine alkaloids<sup>10</sup> such as ant venom, and its application is now in progress.

## Experimental

M.p.s and b.p.s are uncorrected. M.p.s were measured with a Yanaco micro melting point apparatus and b.p.s were measured with a Buchi Kugelrohr apparatus. I.r. spectra were determined

with a JASCO A-102 spectrophotometer. <sup>1</sup>H N.m.r. spectra were run on a JEOL PMX-60 (60 MHz) or JX-270 (270 MHz) spectrometer using CDCl<sub>3</sub> as a solvent. Mass spectra were measured with a JEOL JMS-D200 spectrometer.

General Procedure for Preparation of a-Alkynylazacycloalkanes (10-22).—A mixture of thiolactam (1)-(5) (5 mmol) and iodomethane (8 mmol) in THF (10 ml) was stirred at room temperature for 6 h under argon. The reaction mixture was cooled -78 °C, and to it was added a solution of lithium acetylide (6)-(9), prepared by the dropwise addition of butyllithium (1.6m in hexane, 3.75 ml, 6 mmol) to acetylene (6 mmol) in THF (5 ml) at -78 °C. The mixture was then allowed to warm to -20 °C over 2 h when LiAlH<sub>4</sub> (500 mg, 13 mmol) at the same temperature was added. The mixture was stirred for a further hour at -10 °C when 10% aqueous sodium hydroxide (10 ml) was added to it; the mixture was then extracted with ethyl acetate. The combined organic extracts were washed with brine, dried (MgSO<sub>4</sub>), and evaporated under reduced pressure. Bulb-to-bulb distillation of the crude product using Kugelrhor apparatus yielded the  $\alpha$ -azacycloalkanes (10)—(22) (see Table).

1-(1-*Methylpyrrolidin*-2-*yl*)-2-*phenylacetylene* (**10**). B.p. 72— 79 °C at 0.5 mmHg (Found: C, 82.3; H, 8.1; N, 7.5.  $C_{13}H_{15}N$  requires C, 82.5; H, 8.2; N, 7.6%);  $v_{max}$  (neat) 2 200 and 1 595 cm<sup>-1</sup>;  $\delta_{H}$ (CDCl<sub>3</sub>) 1.47—3.14 (m, 6 H), 2.50 (s, 3 H, NMe), 3.14—3.47 (m, 1 H, NCH), and 7.00—7.57 (m, 5 H, Ph).

1-(1-*Methylpiperidin*-2-*yl*)-2-*phenylacetylene* (11). B.p. 82— 87 °C at 0.5 mmHg (Found: C, 84.2; H, 8.6; N, 6.8. C<sub>14</sub>H<sub>17</sub>N requires C, 84.4; H, 8.6; N, 7.0%);  $v_{max}$  (neat) 2 200 and 1 595 cm<sup>-1</sup>; δ<sub>H</sub>(CDCl<sub>3</sub>) 1.13—2.10 (m, 6 H), 2.10—2.90 (m, 2 H), 2.24 (s, 3 H, NMe), 3.54 (t, *J* 4 Hz, 1 H, NCH), and 7.10—7.60 (m, 5 H, Ph).

1-(1-*Methylperhydroazepin*-2-*yl*)-2-*phenylacetylene* (**12**). B.p. 90—95 °C at 0.5 mmHg (Found: C, 84.8; H, 8.7; N, 6.3. C<sub>15</sub>H<sub>19</sub>N requires C, 84.45; H, 9.0; N, 6.6%);  $v_{max}$ (neat) 2 200 and 1 595 cm<sup>-1</sup>;  $\delta_{H}$ (CDCl<sub>3</sub>) 1.50—2.27 (m, 8 H), 2.27—3.00 (m, 2 H, NCH<sub>2</sub>), 2.50 (s, 3 H, NMe), 3.67—3.97 (m, 1 H, NCH), and 7.13—7.63 (m, 5 H, Ph).

1-(1-*Methylpyrrolidin*-2-*yl*)*hept*-1-*yne* (**13**). B.p. 54—58 °C at 0.6 mmHg (Found: C, 80.0; H, 11.8; N, 7.5.  $C_{12}H_{21}N$  requires C, 80.4; H, 11.8; N, 7.8%);  $v_{max}$ .(neat) 2 220 cm<sup>-1</sup>;  $\delta_{H}$ (CDCl<sub>3</sub>) 0.60 (m, 3 H, Me), 1.03—2.50 (m, 14 H), 2.37 (s, 3 H, NMe), and 2.50—3.57 (m, 1 H, NCH).

1-(1-*Methylpiperidin*-2-*yl*)*hept*-1-*yne* (14). B.p. 59–62 °C at 0.5 mmHg (Found: C, 80.5; H, 12.1; N, 7.0.  $C_{13}H_{23}N$  requires C, 80.8; H, 12.0; N, 7.25%);  $v_{max}$  (neat) 2 225 cm<sup>-1</sup>;  $\delta_{H}$ (CDCl<sub>3</sub>) 0.67–1.04 (m, 3 H, Me), 1.04–2.63 (m, 16 H), 2.37 (s, 3 H, NMe), and 3.00–3.38 (m, 1 H, NCH).

1-(1-*Methylperhydroazepin*-2-*yl*)*hept*-1-*yne* (15). B.p. 68— 70 °C at 0.5 mmHg (Found: C, 79.6; H, 12.15; N, 6.3.  $C_{14}H_{25}N$  requires C, 81.1; H, 12.15; N, 6.8%);  $v_{max}$  (neat) 2 200 cm<sup>-1</sup>;  $\delta_{H}$ (CDCl<sub>3</sub>) 0.60—1.04 (m, 3 H, Me), 1.04—2.80 (m, 18 H), 2.40 (s, 3 H, Me), and 3.33—3.77 (m, 1 H, NCH).

1-(1-*Methylpyrrolydin*-2-*yl*)*prop*-1-*yne* (**16**). B.p. 68—71 °C at 17 mmHg (Found:  $M^+$ , 123.1049. C<sub>8</sub>H<sub>13</sub>N requires M, 123.1047); v<sub>max</sub>.(neat) 2 230 cm<sup>-1</sup>; δ<sub>H</sub>(CDCl<sub>3</sub>) 1.60—2.70 (m, 5 H), 1.85 (d, *J* 2 Hz, 3 H, Me), 2.41 (s, 3 H, NMe), and 2.70—3.30 (m, 1 H, NCH).

1-(1-*Methylpiperidin*-2-*yl*)*prop*-1-*yne* (17). B.p. 64—67 °C at 15 mmHg (Found:  $M^+$ , 137.1202. C<sub>9</sub>H<sub>15</sub>N requires M, 137.1203); v<sub>max</sub>.(neat) 2 230 cm<sup>-1</sup>; δ<sub>H</sub>(CDCl<sub>3</sub>) 1.10—2.00 (m, 6 H), 1.85 (d, *J* 2 Hz, 3 H, CH<sub>2</sub>), 2.00—3.00 (m, 2 H, NCH<sub>2</sub>), 2.83 (s, 3 H, NMe), and 3.00—3.33 (m, 1 H, NCH).

1-(1-*Benzylpyrrolidin*-2-*yl*)-2-*phenylacetylene* (**18**). B.p. 122— 128 °C at 0.4 mmHg (Found: C, 87.5; H, 7.3; N, 5.5.  $C_{19}H_{19}N$  requires C, 87.3; H, 7.3; N, 5.4%);  $v_{max}$  (neat) 2 200 and 1 595 cm<sup>-1</sup>;  $\delta_{H}$ (CDCl<sub>3</sub>) 1.57—3.00 (m, 6 H), 3.60 and 4.08 (ABq, *J* 13 Hz, 2 H, CH<sub>2</sub>Ph), 3.62 (m, 1 H, NCH), and 7.16—7.63 (m, 10 H, 2Ph).

<sup>\* (</sup>  $\pm$  )-Sedaminone has previously been reduced to (  $\pm$  )-sedamine.<sup>9</sup>

1-(1-*Benzylpyrrolidin*-2-*yl*)*hept*-1-*yne* (**19**). B.p. 94—98 °C at 0.5 mmHg (Found: C, 84.8; H, 9.9; N, 5.2. C<sub>18</sub>H<sub>25</sub>N requires C, 84.65; H, 9.9; N, 5.5%);  $v_{max}$  (neat) 2 210 and 1 600 cm<sup>-1</sup>;  $\delta_{\rm H}$ (CDCl<sub>3</sub>) 0.64—1.10 (m, 3 H, Me), 1.10—3.33 (m, 14 H), 3.33—3.50 (m, 1 H, NCH), 3.47, 4.01 (ABq, *J* 13 Hz, 2 H, CH<sub>2</sub>Ph), and 7.30 (s, 5 H, Ph).

1-(1-Benzylpyrrolidin-2-yl)prop-1-yne (**20**). B.p. 73—76 °C at 0.15 mmHg (Found: C, 83.3; H, 8.3; N, 7.1. C<sub>14</sub>H<sub>17</sub>N·0.1H<sub>2</sub>O requires C, 83.6; H, 8.6; N, 7.0%; Found:  $M^+$ , 199.1347. C<sub>14</sub>H<sub>17</sub>N requires M, 199.1359); v<sub>max</sub>.(neat) 2 225 and 1 600 cm<sup>-1</sup>; δ<sub>H</sub>(CDCl<sub>3</sub>) 1.50—3.07 (m, 6 H), 1.87 (d, J 2 Hz, 3 H, Me), 3.07—3.50 (m, 1 H, NCH), 3.45, 4.05 (ABq, J 13 Hz, 2 H, CH<sub>2</sub>Ph), and 7.32 (s, 5 H, Ph).

1-(1-Benzylpiperidin-2-yl)prop-1-yne (**21**). B.p. 88—89 °C at 0.3 mmHg (Found: C, 84.6; H, 8.9; N, 6.5.  $C_{15}H_{19}N$  requires C, 84.45; H, 9.0; N, 6.6%);  $v_{max}$ (neat) 2 225 and 1 600 cm<sup>-1</sup>;  $\delta_{H}$ (CDCl<sub>3</sub>) 1.00—2.10 (m, 6 H), 1.92 (d, *J* 2 Hz, 3 H, Me), 2.10—3.20 (m, 2 H, NCH<sub>2</sub>), 3.20—3.80 (m, 1 H, NCH), 3.51, 3.77 (ABq, *J* 13 Hz, 2 H, CH<sub>2</sub>Ph), and 7.10—7.54 (m, 5 H, Ph).

1-(1-Benzylpyrrolidin-2-yl)-2-trimethylsilylacetylene (**22**). B.p. 86—91 °C at 0.4 mmHg (Found: C, 74.9; H, 9.1; N, 5.6. C<sub>16</sub>H<sub>23</sub>NSi requires C, 74.6; H, 9.0; N, 5.4%); v<sub>max</sub>.(neat) 2 150 and 1 600 cm<sup>-1</sup>; δ<sub>H</sub>(CDCl<sub>3</sub>) 0.22 (s, 9 H, SiMe<sub>3</sub>), 1.53—2.23 (m, 4 H), 2.23—2.87 (m, 2 H, NCH<sub>2</sub>), 3.16—3.20 (m, 1 H, NCH), 3.54, 4.05 (ABq, J 13 Hz, 2 H, CH<sub>2</sub>Ph), and 7.37 (s, 5 H, Ph).

(±)-N-*Methylconiine* (23).—A mixture of (17) (45 mg) and Pd(OH)<sub>2</sub><sup>12</sup> (14 mg) in methanol (3 ml) was stirred under hydrogen (1 atm) for 2.5 h. Filtration followed by washing of the residue with methanol and evaporation of the solvent yielded (23) (58 mg, 100%) as an oil;  $\delta_{\rm H}$ (CDCl<sub>3</sub>) 0.67—1.04 (m, 3 H, Me), 1.04—2.33 (m, 12 H), 2.27 (s, 3 H, NMe), and 2.60—3.10 (m, 1 H, NCH); hydrochloride salt, m.p. 169—170 °C (acetone) (lit.,<sup>12</sup> 168—168.5 °C) (Found: C, 59.9; H, 11.2; N, 7.8. Calc. for C<sub>9</sub>H<sub>20</sub>ClN•0.1H<sub>2</sub>O: C, 60.2; H, 11.25; N, 7.8%).

(±)-Coniine (24).—A mixture of (21) (108 mg) and Pd(OH)<sub>2</sub> (10 mg) in methanol (3 ml) was stirred under hyrogen (1 atm) for 7 h, then worked-up by the method described for (23) to give (24) (81 mg, 100%) as an oil;  $v_{max}$ . 3 300 and 1 650 cm<sup>-1</sup>;  $\delta_{\rm H}$ (CDCl<sub>3</sub>) 0.85 (br t, *J* 6 Hz, 3 H), and 2.00—3.30 (m, 3 H); i.r. and <sup>1</sup>H n.m.r. spectra were identical with those described in the literature.<sup>13</sup> The hydrochloride salt was crystallised from acetone–methanol, m.p. 219—220 °C (lit.,<sup>12</sup> 216—217 °C) (Found: C, 58.6; H, 11.1; N, 8.6. Calc. for C<sub>8</sub>H<sub>18</sub>ClN: C, 58.7; H, 11.1; N, 8.55%).

General Procedure for Monohydroboration and Oxidation of x-Alkynylazacycloalkanes.---To a stirred solution of bis(1,2dimethylpropyl)borane (0.5 mmol), prepared from borane-THF complex (1.0m, 1.5 ml) and 2-methylbut-2-ene (3 mmol, 0.32 ml) by the method of H. C. Brown,<sup>14</sup> was added a solution of a-alkynylazacycloalkane (0.5 mmol) in THF (2 ml) at 0 °C. The mixture was kept in a refrigerator (5 °C) overnight after which it was quenched with 6M aqueous NaOH (0.3 ml) and 30% aqueous  $\hat{H}_2O_2$  (0.6 ml) at 0 °C; it was then stirred for an additional hour at room temperature. The reaction mixture was cooled to 0 °C, 10% hydrochloric acid (2 ml) was added and it was then stirred for 30 min at room temperature and washed with diethyl ether. The aqueous layer was made alkaline with K<sub>2</sub>CO<sub>3</sub> and extracted with diethyl ether. The extract was dried  $(K_2CO_3)$  and concentrated, and the residue separated by alumina column chromatography (hexane-ethyl acetate, 30:1) to give the amino ketones.

(±)-N-*Methylpelletierine* (**25**) (50%). B.p. 67–69 °C at 15 mmHg (Found:  $M^+$ , 155.1288. C<sub>9</sub>H<sub>17</sub>NO requires M, 155.1309); v<sub>max.</sub>(neat) 1 710 cm<sup>-1</sup>;  $\delta_{\rm H}$ (CDCl<sub>3</sub>) 1.19–1.40 (m, 2 H), 1.47–1.68 (m, 4 H), 2.18 (s, 3 H, Me), 2.21 (s, 3 H, NMe),

2.11–2.45 (m, 2 H), 2.52–2.59 (m, 1 H), and 2.75–2.83 (m, 2 H). The hydrochloride salt was crystallised from ethyl acetate, m.p. 161.5–162.5 °C (lit.,<sup>15</sup> 158–159 °C) (Found: C, 56.4; H, 9.4; N, 7.3. Calc. for  $C_9H_{18}CINO$ : C, 56.4; H, 9.5; N, 7.3%).

*Ethyl* 1-*methylpiperidin*-2-*yl* ketone (**26**) (14%). B.p. 65— 67 °C at 15 mmHg (Found:  $M^+$ , 155.1303. C<sub>9</sub>H<sub>17</sub>NO requires M, 155.1305); v<sub>max</sub>.(neat) 1 710 cm<sup>-1</sup>;  $\delta_{\rm H}$ (CDCl<sub>3</sub>) 1.06 (t, J 7 Hz, 3 H, Me), 1.18—2.69 (m, 10 H), 2.13 (s, 3 H, NMe), and 2.90— 2.95 (m, 1 H, NCH).

(±)-Sedaminone (28). 50%; b.p. 92–93 °C at 0.15 mmHg (lit.,<sup>16</sup> 135–136 °C at 1 mmHg) (Found:  $M^+$ , 217.1437. Calc. for C<sub>14</sub>H<sub>19</sub>NO: M, 217.1465); v<sub>max</sub>.(neat) 1 680 and 1 595 cm<sup>-1</sup>;  $\delta_{\rm H}$ (CDCl<sub>3</sub>) 1.00–2.53 (m, 7 H), 2.32 (s, 3 H, NMe), 2.53–3.92 (m, 4 H), 7.30–8.20 (m, 5 H, Ph). Picrate, m.p. 159.5–161 °C (lit.,<sup>16</sup> 159–160 °C) (Found: C, 53.7; H, 5.0; N, 12.75. Calc. for C<sub>20</sub>H<sub>22</sub>N<sub>4</sub>O<sub>8</sub>: C, 53.8; H, 5.0; N, 12.55%).

*Benzyl* 1-*methylpiperidin*-2-*yl* ketone (**29**) (18%). B.p. 87— 90 °C at 0.15 mmHg (Found: C, 77.0; H, 8.85; N, 6.6.  $C_{14}H_{19}NO$  requires C, 77.4; H, 8.8; N, 6.45%);  $v_{max}$  (neat) 1 715 and 1 600 cm<sup>-1</sup>;  $\delta_{H}(CDCl_{3})$  0.90—2.33 (m, 6 H), 2.13 (s, 3 H, NMe), 2.47—3.27 (m, 3 H, NCH and NCH<sub>2</sub>), 3.83 (s, 2 H, CH<sub>2</sub>Ph), and 7.34 (s, 5 H, Ph).

(±)-*Hygrine* (**30**) (54%). B.p. 71.5—75 °C at 15 mmHg;  $v_{max}$ .(neat) 1 710 cm<sup>-1</sup>;  $\delta_{H}$ (CDCl<sub>3</sub>) 1.00—2.73 (m, 6 H), 2.13 (s, 3 H), 2.26 (s, 3 H), and 2.73—3.53 (m, 3 H). I.r. and <sup>1</sup>H n.m.r. spectra were identical with those described in the literature.<sup>17</sup> Picrate, m.p. 151—153 °C (ethanol) (lit.,<sup>18</sup> 153.5—155 °C) (Found: C, 45.4; H, 4.9; N, 15.4. Calc. for C<sub>14</sub>H<sub>18</sub>N<sub>4</sub>O<sub>8</sub>: C, 45.4; H, 4.9; N, 15.1%).

(±)-N-*Benzylhygrine* (**31**) (52%). B.p. 87.5–91.5 °C at 0.15 mmHg (Found: C, 77.6; H, 8.7; N, 6.2.  $C_{14}H_{19}NO$  requires C, 77.4; H, 8.8; N, 6.45%);  $v_{max}$  (neat) 1 710 and 1 600 cm<sup>-1</sup>;  $\delta_{H}$ (CDCl<sub>3</sub>) 1.00–3.20 (m, 9 H), 2.17 (s, 3 H, Me), 3.42, 3.99 (ABq, *J* 13 Hz, 2 H, CH<sub>2</sub>Ph), and 7.38 (s, 5 H, Ph).

## Acknowledgements

We gratefully acknowledge the partial financial assistance of The Research Foundation for Pharmaceutical Sciences for this study.

## References

- (a) K. Shiosaki and H. Rapoport, J. Org. Chem., 1985, 50, 1229; (b)
   P. Q. Huang, S. Arseniyadis, and H.-P. Husson, Tetrahedron Lett., 1987, 28, 547; (c) M. Yamaguchi and I. Hirao, *ibid.*, 1983, 23, 1719.
- 2 For reviews see (a) H. Takahata and T. Yamazaki, J. Syn. Org. Chem. Jpn., 1987, **45**, 686; (b) Heterocycles, 1988, **27**, 1953.
- 3 Y. Tominaga, S. Kohra, and A. Hosomi, *Tetrahedron Lett.*, 1987, 28, 1529.
- 4 H. Takahata, E.-C. Wang, T. Nakajima, and T. Yamazaki, *Chem. Pharm. Bull.*, 1987, **35**, 3139.
- 5 T. Yamaguchi, Y. Shimizu, and T. Suzuki, Chem. Ind., 1972, 380.
- 6 K. Yoshida, S. Nakajima, T. Wakamatsu, Y. Ban, and M. Shibasaki, *Heterocycles*, 1988, 27, 1167.
- 7 The hydroboration of olefinic amines has been reported: (a) A. Dicko, M. Montury, and M. Baboulene, *Tetrahedron Lett.*, 1987, 28, 6041; (b) M. M. B. Nemia, J. Lee, and M. M. Joullie, *Synth. Commun.*, 1983, 13, 1117; (c) M. Baboulene, J.-L. Torregrosa, V. Speziale, and A. Lattes, *Bull. Soc. Chim. Fr.*, 1980, 565.
- 8 G. Zweifel, G. M. Clark, and N. Polston, J. Am. Chem. Soc., 1971, 93, 3395.
- 9 H. C. Beyerman, W. Eveeleens, and Y. M. F. Muller, *Recl. Trav. Chem.*, 1956, **75**, 63.
- 10 (a) G. Strunz and J. Findlay, in 'The Alkaloids,' A. Brossi, (ed.), Academic Press, New York, 1986, vol. 26, pp. 89–183; (b) G. Massiot and C. Delaude, *ibid.*, 1986, vol. 27, pp. 269–322; (c) A. Numata and T. Ibuka, *ibid.*, 1987, vol. 31, pp. 193–315.
- 11 S. Nishimura, M. Shimahara, and M. Siota, J. Org. Chem., 1966, 31, 2394.

- 12 'Dictionary of Organic Compounds,' J. Buckingham (ed.), Chapman and Hall, New York, 1982, vol. 5, p. 4800-4801.
- 13 K. Aketa, S. Terashima, and S. Yamada, Chem. Pharm. Bull., 1976, **24**, 621.
- 14 H. C. Brown, 'Organic Synthesis via Boranes,' Wiley International, New York, 1975, p. 29—31.
  15 E. Leistner and I. D. Spenser, J. Am. Chem. Soc., 1973, 95, 4715.
- 16 H. C. Beyerman and P. H. Enthoven, Recl. Trav. Chim. Pays-Bas, 1956, **75**, 82.
- 17 I. R. C. Bick, J. W. Gillard, and H. M. Leow, Aust. J. Chem., 1979, 32, 2523.
- 18 R. Ghirlando, A. S. Howard, R. B. Katz, and J. P. Michael, Tetrahedron, 1984, 40, 2879.

Received 4th October 1988; Paper 8/03953D